



Enzymatic Assay of α -GALACTOSIDASE (CDLS-1)

This product is for research use only and is not intended for diagnostic use.

PRODUCT INFORMATION

Size 5 Unit

Principles of Testing

PNP α -D-Galactopyranoside + H₂O $\xrightarrow{\alpha\text{-Galactosidase}}$ p-Nitrophenol + D-Galactose

Abbreviation used:

PNP α -D-Galactopyranoside = p-Nitrophenyl α -D-Galactopyranoside

Reagents And Materials Provided

- A. 100 mM Potassium Phosphate Monobasic Solution
(Prepare 100 ml in deionized water using Potassium Phosphate, Monobasic, Anhydrous)
- B. 100 mM Potassium Phosphate Dibasic Solution
(Prepare 100 ml in deionized water using Potassium Phosphate, Dibasic, Trihydrate,)
- C. 100 mM Potassium Phosphate Buffer, pH 6.5 at 25°C
(Prepare 100 ml by adjusting 50 ml of Reagent A to pH 6.5 at 25°C by adding Reagent B.)
- D. 9.9 mM p-Nitrophenyl α -D-Galactopyranoside Solution (PNP-Gal)
(Prepare 4 ml in deionized water using p-Nitrophenyl α -D-Galactopyranoside.)
- E. 200 mM Borate Buffer, pH 9.8 at 25°C
(Prepare 100 ml in deionized water using Boric Acid, Sigma Prod. No. B-0252. Adjust to pH 9.8 at 25°C with 1 M NaOH.)
- F. α -Galactosidase Enzyme Solution
(Immediately before use, prepare a solution containing 0.05 - 0.10 units/ml of α -Galactosidase in cold Reagent C.)

Assay Procedure

Pipette (in milliliters) the following reagents into suitable cuvettes:

	Test	Blank
Reagent C (Potassium Phosphate Buffer)	0.70	0.70
Reagent D (PNP-Gal)	0.20	0.20

Mix by swirling and equilibrate to 25°C. Then add:

Reagent F (Enzyme Solution)	0.10	-----
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Immediately mix by swirling and incubate at 25°C for exactly 5 minutes. Then add:

Reagent E (Borate Buffer)	2.00	2.00
Reagent F (Enzyme Solution)	-----	0.10

Mix by swirling and record the A405nm

for both the Test and Blank, using a suitably thermostatted spectrophotometer.

Calculation

$$\text{Units/ml enzyme} = \frac{(\text{A405nm Test} - \text{A405nm Blank})(3.0)(\text{df})}{(18.5)(5)(0.1)}$$

3.0 = Total volume of assay

df = Dilution factor

5 = Conversion factor for 5 minutes to 1 minute

18.5 = Millimolar extinction coefficient of p-Nitrophenol at 405 nm

0.1 = Volume (in milliliter) of enzyme used

$$\text{Units/mg solid} = \frac{\text{units/ml enzyme}}{\text{mg solid/ml enzyme}}$$

$$\text{Units/mg protein} = \frac{\text{units/ml enzyme}}{\text{mg protein/ml enzyme}}$$

References

(1968) Eur. J. Biochem. 8, 395
